## IMPROVED SYNTHETIC ACCESS TO THE B, Y-ENOL ETHER ANINO ACIDS, L-2-AMINO-4-METHOXY-<u>TRANS</u>-BUT-3-ENOIC ACID AND L-2-AMINO-4-METHOXY-<u>CIS</u>-BUT-3-ENOIC ACID

VITAUTS ALKS AND JANICE R. SUFRIN\* Grace Cancer Drug Center Roswell Park Cancer Institute Buffalo, NY 14263, USA

Summary: A synthetic route to L-2-amino-4-methoxy-<u>trans</u>-but-3-enoic acid, exclusive of the <u>cis</u>isomer, has been developed. The key step is direct formation of a <u>trans</u>-enol ether derivative from the corresponding dimethylacetal, by refluxing in CCl<sub>4</sub>, in the presence of hexamethyldisilazane and Me<sub>3</sub>SiI. The isomeric L-<u>cis</u> amino acid could be accessed from this route by isomerizing the intermediate, methyl DL-2-acetamido-2-amino-4-methoxy-<u>trans</u>-but-3-enoate at 217° C to give a <u>cis</u>-<u>trans</u> mixture in a 1:10 ratio.

Dehydroamino acids are of particular biological interest for their properties as enzyme inhibitors,<sup>1,2</sup> as well as components of peptide structures.<sup>3</sup> The specific class of  $B,\gamma$ -enol ether amino acids has presented synthetic challenges arising from the need to generate an enol ether moiety in the presence of potentially reactive amino acid substituents and to deprotect the latter groups without degrading the acid sensitive enol ether substituent.

The isomeric amino acids, L-2-amino-4-methoxy-<u>trans</u>-but-3-enoic acid<sup>4</sup> (L-transAMB) and L-2amino-4-methoxy-<u>cis</u>-but-3-enoic acid<sup>2</sup> (L-cisAMB) are representatives of this class, which also includes the naturally occurring amino acids L-2-amino-4-(2-aminomethoxy-<u>trans</u>-but-3-enoic acid<sup>5</sup> and rhizobitoxine.<sup>6</sup> Synthesis of this latter amino acid has relied on the use of a protected derivative of L-transAMB as a key reaction intermediate. L-transAMB was originally isolated from <u>Pseudomonas aeruginosa</u>.<sup>4</sup> Extensive biological characterization has shown it to be a potent inhibitor of ethylene biosynthesis in plants<sup>7</sup> and a potent inhibitor of cultured Walker 256 carcinosarcoma cells.<sup>8</sup> It inhibits several enzymes, including serine hydroxymethyltransferase,<sup>9</sup> porphobilinogen synthase,<sup>10</sup>  $\delta$ -aminolevulinic acid synthetase,<sup>11</sup>  $\delta$ -aminolevulinic acid dehydrase<sup>11</sup> and L-aspartate amino transferase.<sup>1</sup> It was demonstrated to be a potent irreversible k<sub>cat</sub> inhibitor of the latter enzyme, suggesting that other  $\beta,\gamma$ -unsaturated amino acids might similarly inhibit related pyridoxal-linked enzymes. 5258

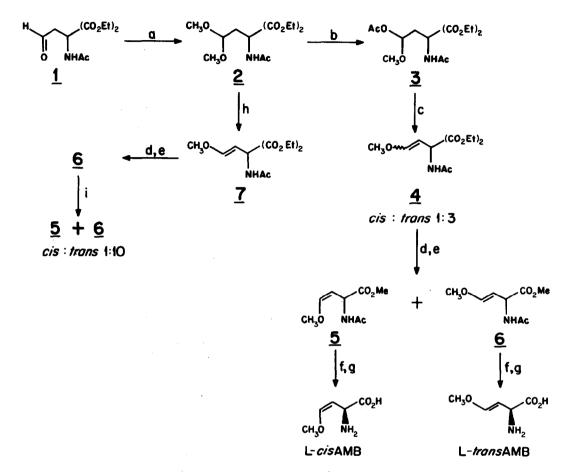
L-cisAMB, originally obtained as a minor component of L-transAMB synthesis,<sup>2,12</sup> has also proved to be of considerable biological interest. It is a potent methionine analog inhibitor of S-adenosylmethionine (AdoMet) synthetase<sup>2</sup> and has been used to study the biochemical consequences of AdoMet depletion in cells.<sup>13,14,15</sup> Its antitumor activity has also been evaluated.<sup>16</sup>

The original preparation of these two isomeric amino acids utilized a synthetic route in which the key step was pyrolysis of a labile hemiacetal ester derivative 3, to give a cis-trans mixture of 4, which, after decarbethoxylation, gave 5 and 6 in a 1:3 <u>cis-trans</u> ratio.<sup>2,12</sup> Chromatographic separation, followed by enzymatic resolution, has consistently provided L-transAMB and L-cisAMB in respective overall yields of 9.5% and 1.6% in our scale-up preparations. More recently, syntheses of  $5^{17}$  and  $5^{18}$  exclusive of each other, have been reported. Both utilize a convergent strategy to obtain the desired <u>cis</u>- or <u>trans</u>-intermediate, although in the case of **6**, the key step is the isomerisation of an  $\alpha$ , B-dehydroamino acid derivative to give the <u>trans</u>-isomer. The new route to L-transAMB<sup>18</sup> provides versatility with regard to the stereoselectively. introduction of different enol ether groups and N-protective groups, but requires at least nine synthetic steps to obtain the penultimate intermediate  $\underline{6}$ . The new route to  $\underline{5}$  provides a means to obtain L-cisAMB without the necessity of tedious chromatographic separations of protected cistrans intermediates, but significant improvements in the final yield of the L-amino acid were not realized because of a low yield in the key condensation step.

Novel methods to improve synthetic access to L-cisAMB and L-transAMB remain of interest from both a chemical and a biological viewpoint. In this regard, we have developed modifications of the synthesis by Keith et al.,<sup>12</sup> which make this original synthetic strategy more attractive for obtaining either of these isomeric amino acids: one modification provides N-Ac-DL-transAMB methyl ester  $\underline{6}$ , stereoselectively; the second modification provides access to additional quantities of N-Ac-DL-cisAMB methyl ester 5 from the pure trans-isomer, 6. As shown in the accompanying synthetic scheme,  $\underline{Z}$  could be obtained directly from the dimethyl acetal  $\underline{2}$  (2.3 g) in 77% yield, by refluxing in CCl, with trimethylsilyl iodide (Me<sub>s</sub>SiI, 1.2 equiv) and hexamethyldisilazane (HMDS, 2.5 equiv) for 4 hr, a procedure adapted from Miller and McKean's general method for facile preparation of methyl enol ethers from acetals.<sup>19</sup> In addition, we were able to isomerize pure 6 by heating at  $217^{\circ}$  C (for 1.5 hr under an argon atmosphere), to give a mixture of 5 and 6 in a cistrans ratio of 1:10. Since pure 6 could be recovered (72%), significant quantities of 5 were accumulated for the final resolution step by repetitive isomerization and chromatography, albeit in a tedious manner. These two modifications provide easier access to both L-cisAMB and/or LtransAMB and as a result, enhance the versatility of the original synthetic route which can more readily accommodate the need for increased amounts of either isomeric amino acid.

## Acknowledgement

This research was supported by Grants CA13038 and CA24538 from the National Cancer Institute.



(a)  $(MeO)_3CH$ ,  $NH_4Cl$ , reflux (b)  $Ac_2O$ , Dowex 50-X8 (H<sup>+</sup>), 65° C (c) 185° C, 17mm (d) NaOMe, MeOH, RT (e) chromatography (f) aq. LiOH (g) hog kidney acylase I, 40° C, pH 7.4, 16 h (h) Me<sub>3</sub>SiI, HMDS, CCl<sub>4</sub>, reflux, 4 h (i) 217° C, argon, 1.5 h.

## References

- 1. R. R. Rando, Nature, 1974, 250, 586.
- 2. J. R. Sufrin, J. B. Lombardini and D. D. Keith, Biochem. Biophys. Res. Commun. 1982, 106, 551.
- 3. U. Schmidt, A. Liebernecht and J. Wild, Synthesis 1988, 159.
- 4. U. Sahm, G. Knobloch and F. Wagner, J. Antibiot. 1973, 26, 389.
- 5. D. D. Keith, R. Yang, J. A. Tortora and M. Weigele, J. Org. Chem. 1978, 43, 3713.
- 6. D. D. Keith, J. A. Tortora, K. Ineichen and W. Leimgruber, Tetrahedron, 1975, 31, 2633.
- 7. A. K. Matoo, J. D. Anderson, E. Chalutz and M. Lieberman, Plant Physiol. 1979, 64, 289.

- 8. M. J. Tisdale, Biochem. Pharmacol. 1980, 29, 501.
- 9. M. J. Tisdale, Chem. Biol. Interactions, 1981, 34, 75.
- 10. T. Dashman, Life Sci., 1980, 27, 1415.
- 11. T. Dashman and J. J. Kamm, Life Sci. 1979, 24, 185.
- 12. D. D. Keith, J. A. Tortora and R. Yang, J. Org. Chem. 1978, 43, 3711.
- 13. D. L. Kramer, J. R. Sufrin and C. W. Porter, Biochem. J. 1987, 247, 259.
- 14. D. L. Kramer, J. R. Sufrin and C. W. Porter, Biochem. J. 1988, 249, 581.
- 15. D. L. Kramer, C. W. Porter, R. T. Borchardt and J. R. Sufrin, Cancer Res. 1990, in press.
- J. R. Sufrin, J. B. Lombardini, D. L. Kramer, V. Alks, R. J. Bernacki and C. W. Porter, in Biological Nethylation and Drug Design. R.T. Borchardt, C. R. Creveling and P. M. Ueland, Eds., Humana Press, Clifton, New Jersey, 1986, p 373.
- 17. V. Alks and J. R. Sufrin, Syn. Commun. 1989, 19, 1479.
- 18. M. Daumas, L. Vo-Quang, Y. Vo-Quang and F. Le Goffic, Tet. Lett. 1989, 30, 5121.
- 19. R. D. Miller and D. R. McKean, Tet. Lett. 1982, 23, 323.

(Received in USA 3 July 1990)